

ASTR/GEOL-2040: Search for life in the Universe: Lecture 27

- Titan
- Methan cycle
- How is it sustained?

Axel Brandenburg

(Office hours: Mondays 2:30 – 3:30 in X590 and

Wednesdays 11-12 in D230)

Visitor today

- Emilie Royer (LASP, Laboratory for Atmospheric & Space Physics)
- Worked on upper atmospheres of most of the bodies of the solar system (Venus, Mars, Titan, Enceladus, and many more)

Moons of Jupiter & Saturn: a comparison

Jupiter	orbit	Saturn	orbit
Io	1.8 d	Enceladus	1.4 d
Europa	3.6 d		
Ganymede	7.2 d		
Callisto	17 d	Titan	16 d

Overview

- Saturn: 62 moons; Titan the only one with dense atmosphere
- Exotic environment, but no evidence for life (yet?)
- plays important role for our understanding
 - Photochemical processes important!
 - Wide range of organic molecules
 - A bit like early Earth?

Historical background

- Discovered by Huygens (1629 – 1695)
- Named in mid 1800 by John Herschel (son of William)
 - After Saturn’s mythical brothers
- Limb darkening (Comas Sola, 1907) → atmosphere
- Gerard Kuiper (1940) with 82 inch (2m) McDonald Observatory: methane
- First “visited” by Pioneer 11 (Sept 1979): 363,000 km, Voyager 1 (Nov 1980): 4394 km



TITAN: A SATELLITE WITH AN ATMOSPHERE*

GERARD P. KUIPER¹

McDonald and Yerkes Observatories

Received August 21, 1944

ABSTRACT

Recently the ten largest satellites in the solar system, as well as Pluto, were observed spectroscopically. Only Titan was found to have an atmosphere of sufficient prominence to be detected, but Triton and Pluto require further study. The composition of Titan's atmosphere is similar to that of Saturn, although the optical thickness is somewhat less.

The presence of gases rich in hydrogen atoms on a small body like Titan is surprising and indicates that the atmosphere was formed after Titan had cooled off. Similar arguments, though less compelling, may be advanced for analogous conclusions in regard to the formation of the atmospheres of Mars, Venus, and the earth.



Titan, before 2003



Voyager 1981

- Second-largest moon in SS
- Density $\sim 1900 \text{ kg/m}^3$
- thick atmosphere! Sublimation
 - N_2 (96%), methane, ethane
 - ethane may condense
 - clouds, rain, oceans/lakes/ponds/erosion
- Geology was unknown due to thick atmosphere

Fly-bys

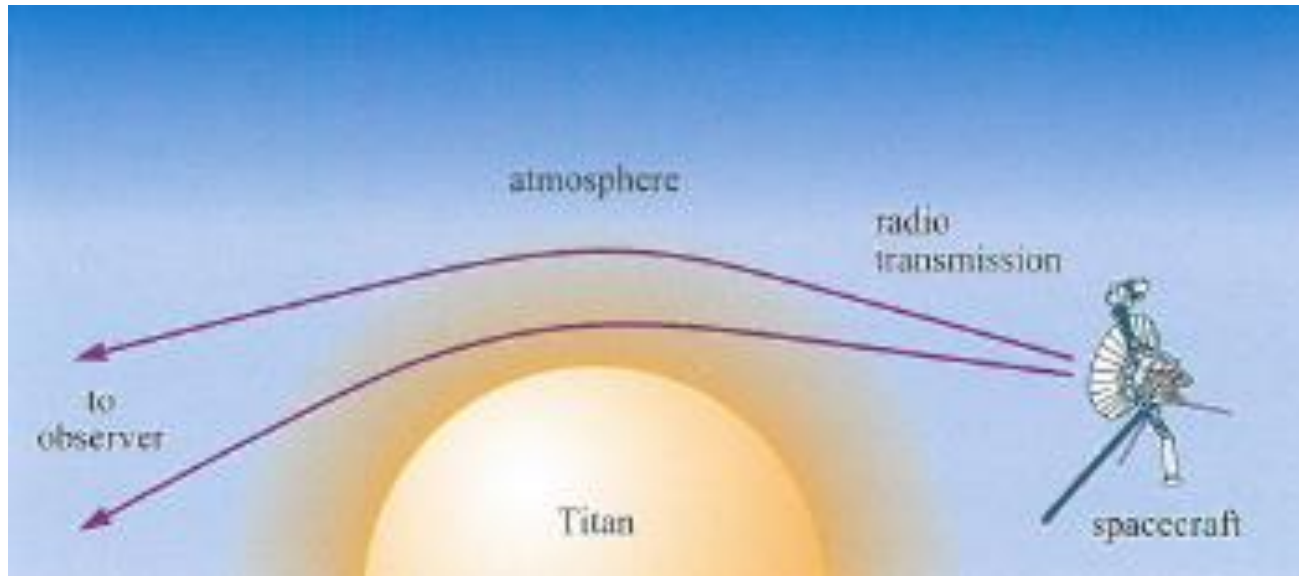
- Not limited by Earth atmosphere
- Infrared, ultraviolet
- Spatial resolution
- Higher intensity of radiation from object
-
- ...

Voyager 1

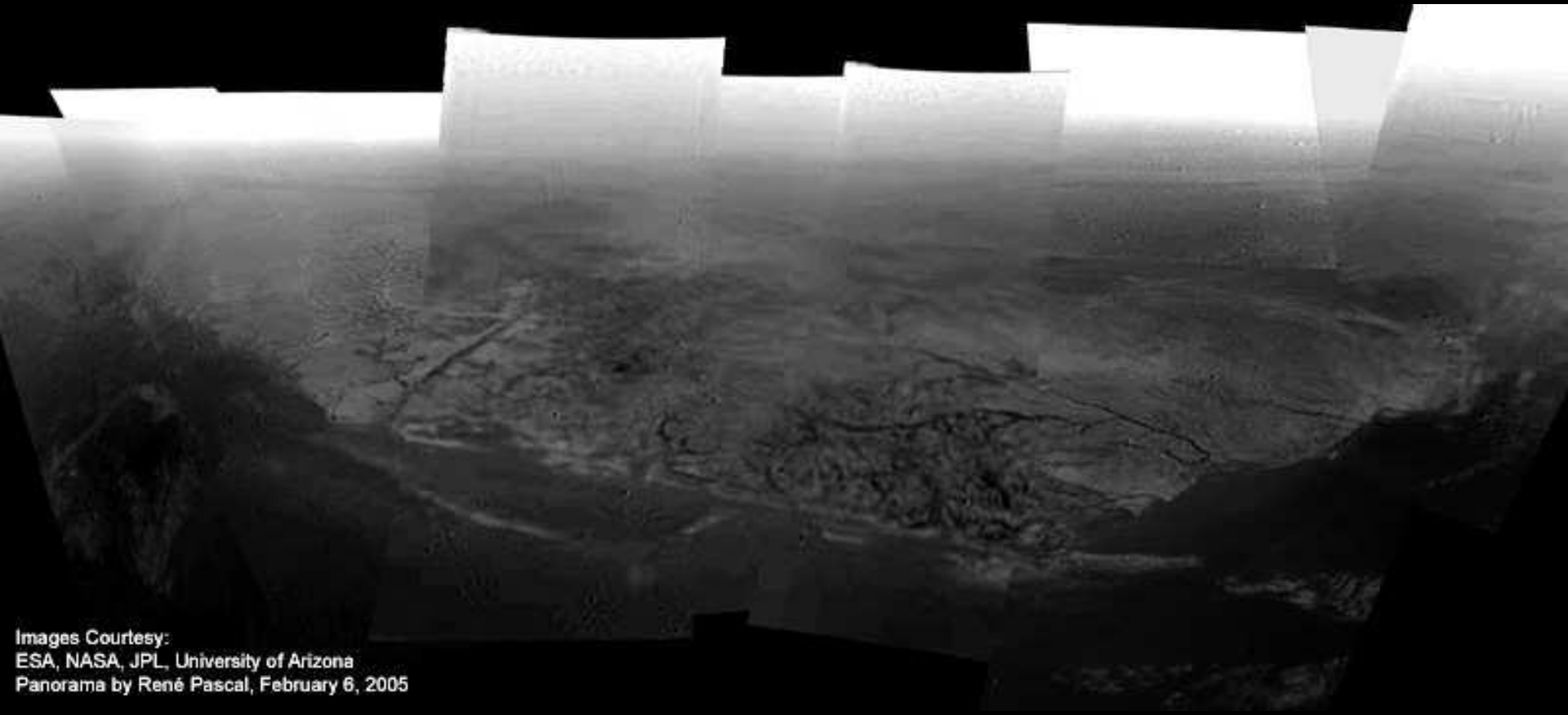
- Orange haze
 - But featureless
- Dark band near near north pole
 - Slight contrast between north/south
- No gaps in haze, no clouds
 - No wind velocity measurements
- Complex spectra

Advantages of fly-bys...

- Measuring thickness of atmosphere
- 1.6 bar

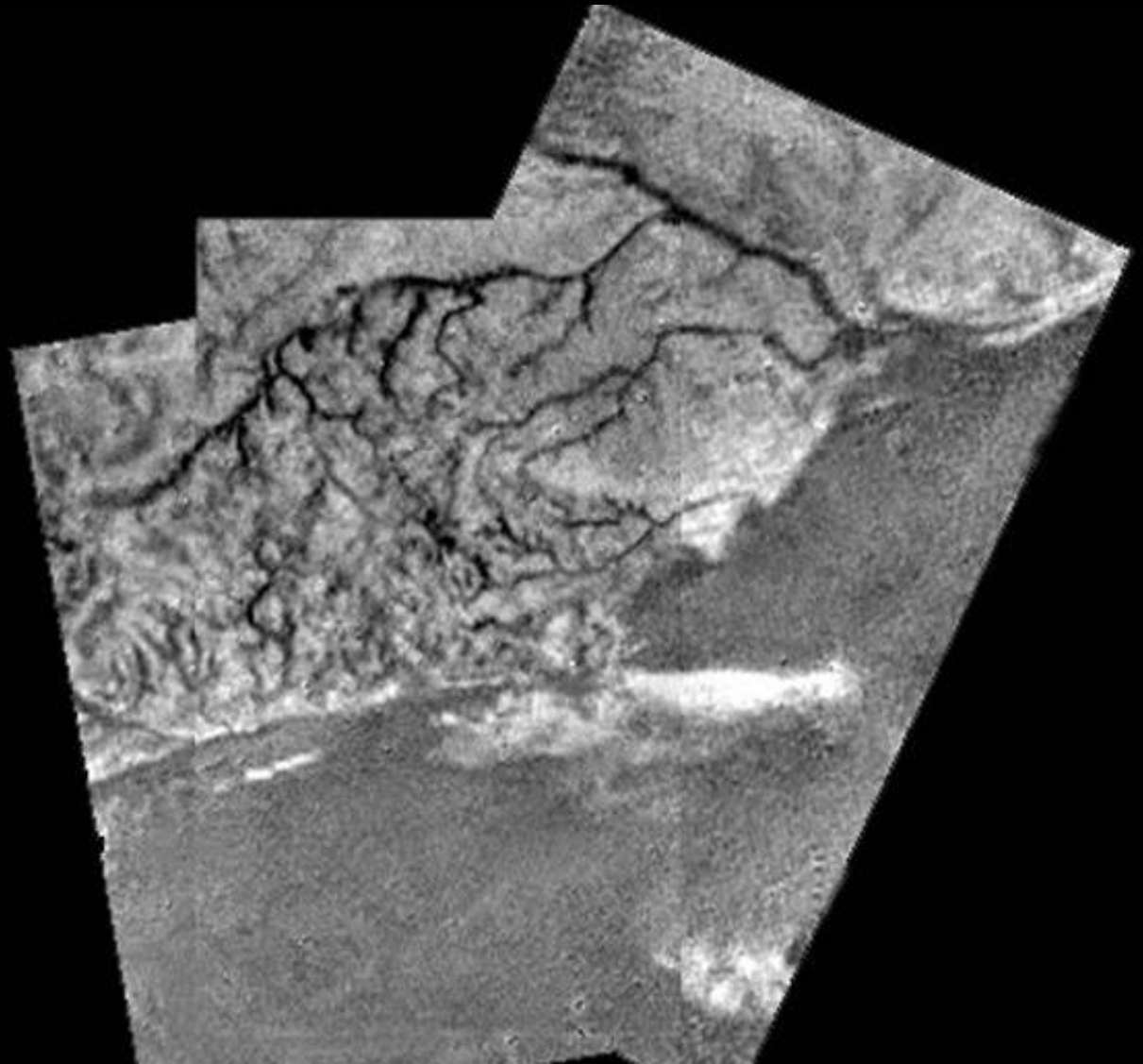
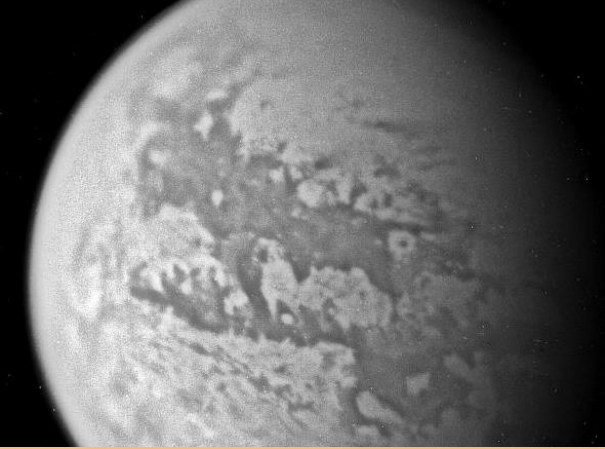


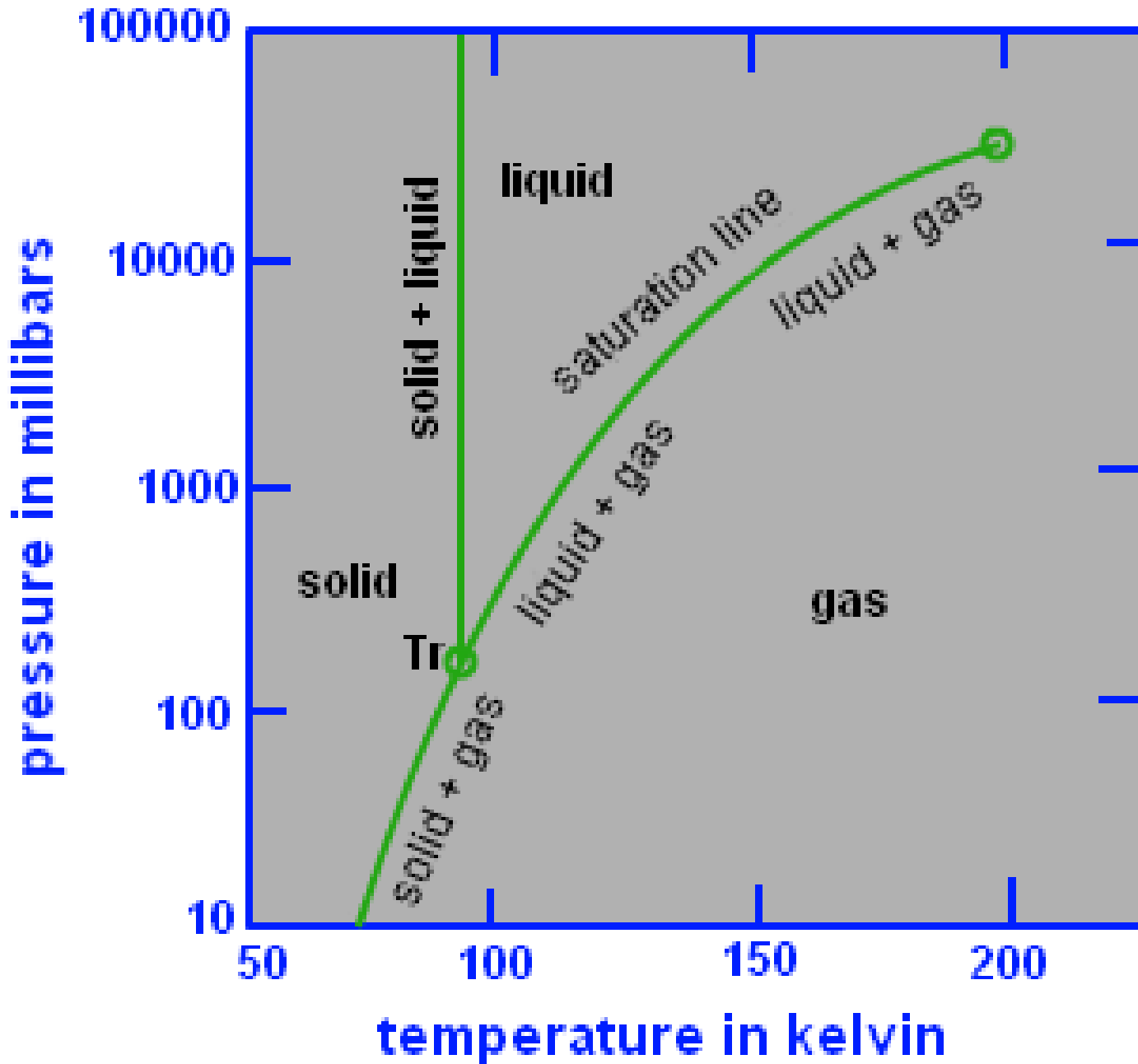
Huygens descent, Jan 14, 2015



Images Courtesy:
ESA, NASA, JPL, University of Arizona
Panorama by René Pascal, February 6, 2005

Lakes, rivers, riverbeds





Methane phase diagram

- At triple point of methane
- What does this imply?

solid, liquid, gas
can all exist

EARTH

km

50 280K

NITROGEN
OXYGEN
ARGON

30 240K

OZONE

20

210K

10

290K

WATER



TITAN

600
km 160K

NITROGEN
METHANE
ARGON(?)

500

400

300

THIN HAZE LAYER

200

THICK
PHOTOCHEMICAL HA
PARTICULATE
RAIN?

160K

100

120K

72K

95K

METHANE

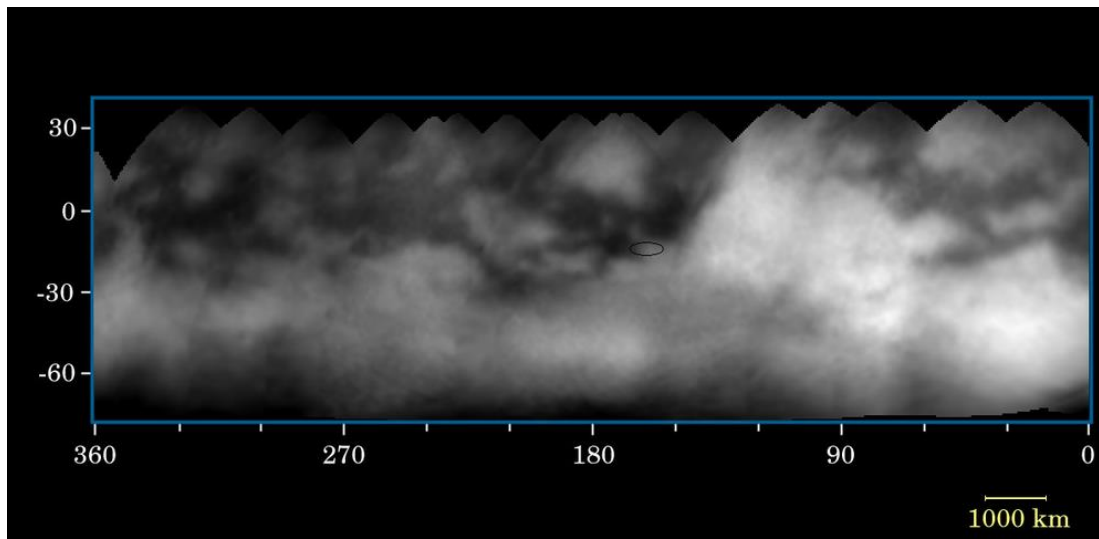


Ice versus terrestrial

	Rocky planet	Icy body
Core	Iron	Rock
Mantle	Silicate rocky shell	Liquid water
Crust	Rock	Water ice
Outgassing	CO ₂	CH ₄
Eruption of ...	Lava	Slush
Aerosols	Dust	Haze
Dunes	Sand	Organics
Cycles	Water	Methane

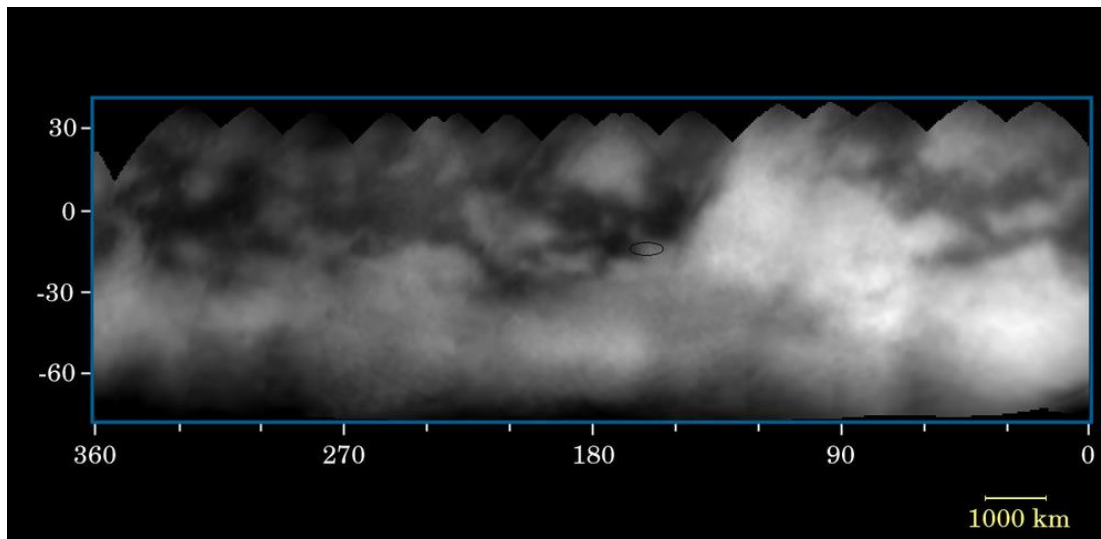
Surface topography

- Rivers from bright highlands
- To dark/flat lowlands
- Coast lines
- Pebbles: round (why?)



Stones made of water ice!

- Ice balls deposited by powerful currents
- Liquid methane obvious possibility



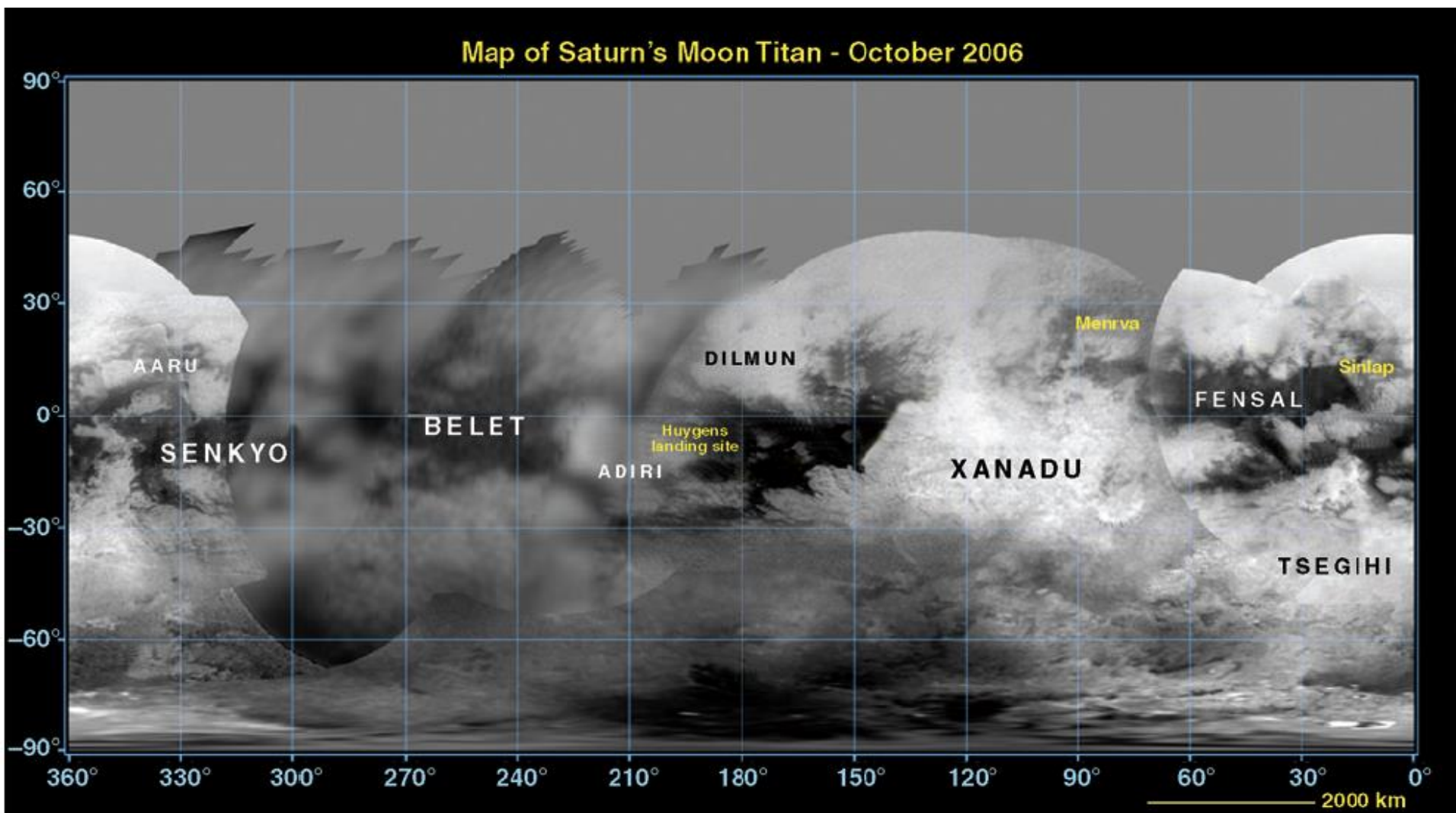


Figure 3. A map of Titan from Cassini ISS images at 940 nm, with names indicated for major regions (names are selected by a committee of the IAU). Some areas are mapped at rather higher resolution than others; latitudes 50°N and higher are insufficiently illuminated in the season shown to map. (Image courtesy of NASA/JPL/Ciclops.)

Precipitation in the Summer

- A. More precipitation
- B. Less precipitation
- C. No difference

Precipitation in the Summer

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Evaporation from tropics!

autumn. There is evidence for rainfall at high latitudes in summer. It seems that methane evaporates from the warm tropics over summer and is transported toward the poles where it falls as rain. This recharges the lakes but leaves the tropics a dry desert

Longstaff, p.293

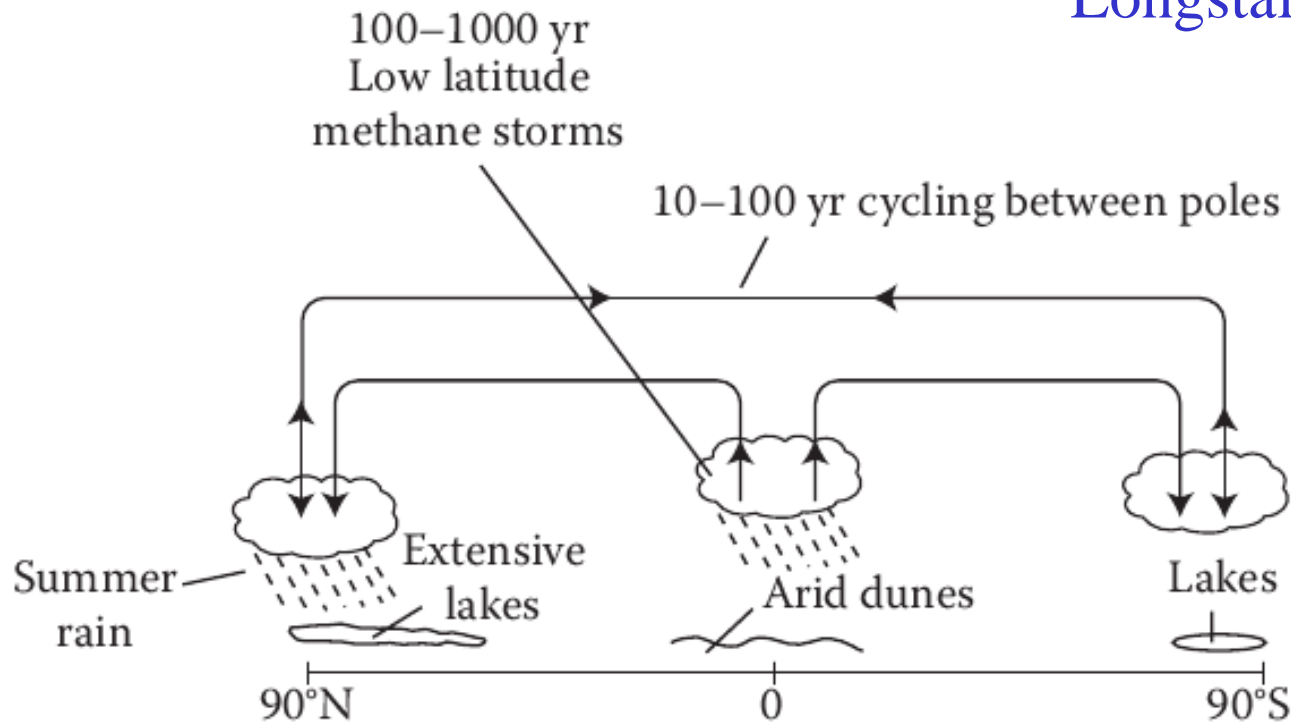
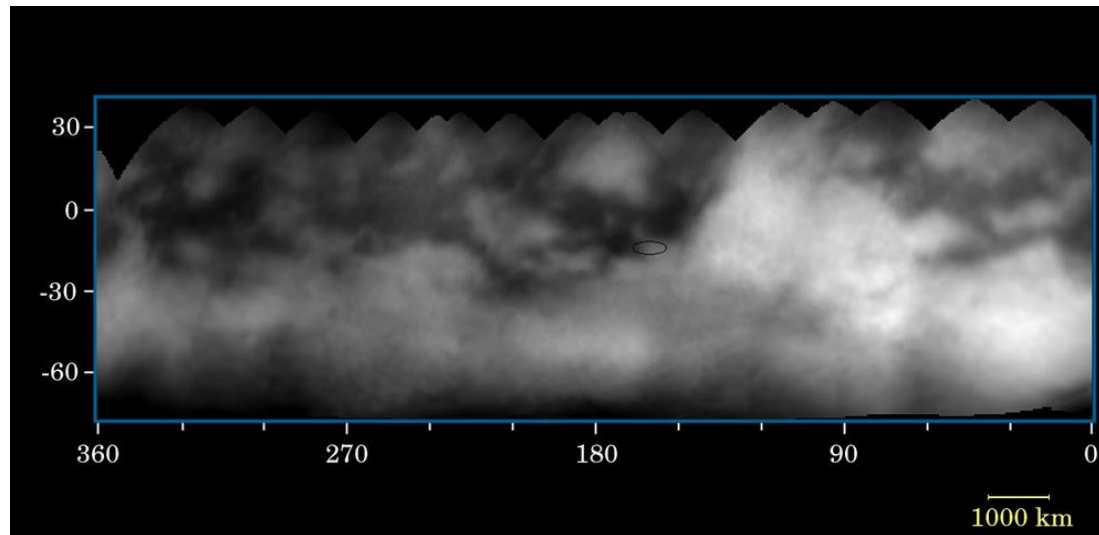
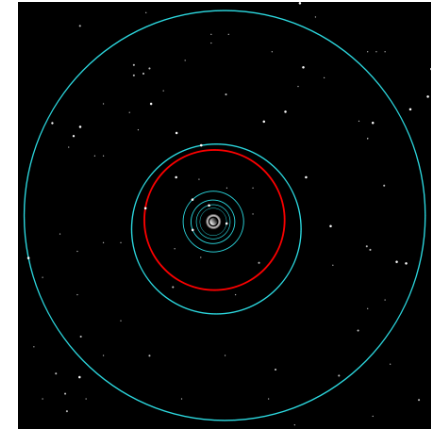


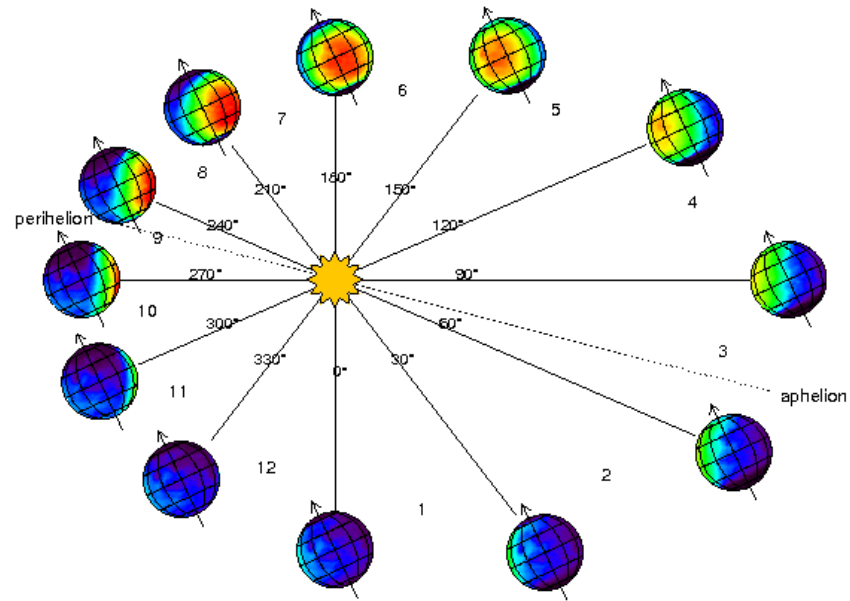
FIGURE 11.2 The methane cycle on Titan evaporates methane in equatorial regions in summer and transports it to polar regions where it rains.

Northern summer stronger

- Because of eccentricity
- Longer rainy season than in the south
- More lakes in the north

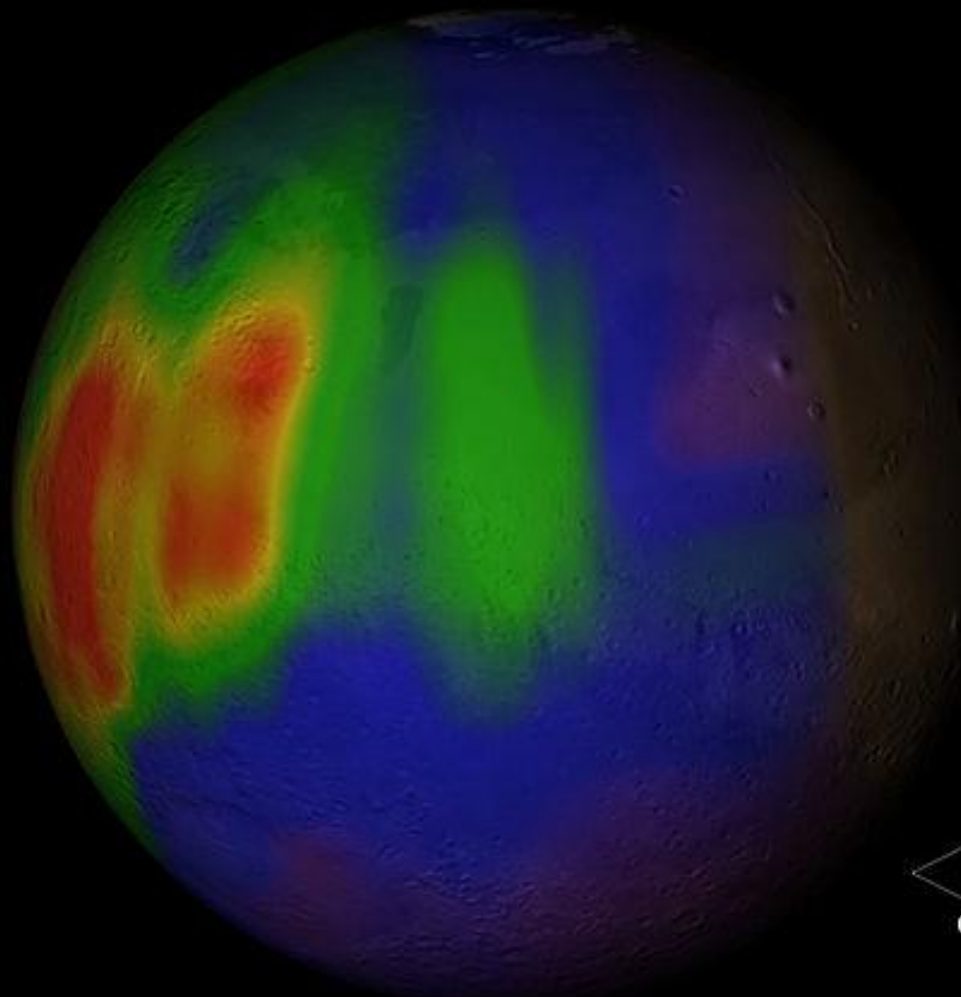


Similar to Martian seasons



	Summer	Winter
North	Cooler	wamer
	longer	shorter
	caps sublimate less	Caps freeze less
South	hotter	cooler
	shorter	longer
	caps sublimate more	freeze more

Comparison with Mars: constant loss of methane



Methane release:
Northern summer



Lake reservoir sufficient?

- Loss ~ 100 kg/s
 - Exercise 5.4 in RGS, p.195
- If 10% of surface methane lakes
 - 400 m deep on average
 - Total 1.5×10^{18} kg
- Time: 1.5×10^{18} kg / (100 kg/s)
 - 1.5×10^{16} s = 0.5 Gyr

Main results so far

- Titan is the size of a planet, and resembles the Earth, with a thick, organic-rich and complex atmosphere
- Methane-ethane lakes resupply the methane in the atmosphere, which is continuously released from Titan's interior by volcanoes
- Titan shows pre-biotic chemistry, like on Earth before life arose
- Life using methane as a solvent or eating acetylene are possible

Life on Titan?

- If not, it resembles Earth before life: plenty of organics and chemical reactions: Can organisms eat these?
- After giant impacts? ‘Titan Spring’
- Hot springs?
- At great depths: warm pond or deep-sea vent?
- Methane-based or acetylene eating life?

Next time

- Titan's subsurface ocean habitable?
- Methanogenesis, cryovolcanism
- Haze photochemistry

- Longstaff: pp 297 – 303
- BS: 319 – 326
- RGS: 180 – 198