

ASTR/GEOL-2040: Search for life in the Universe, Lecture 7

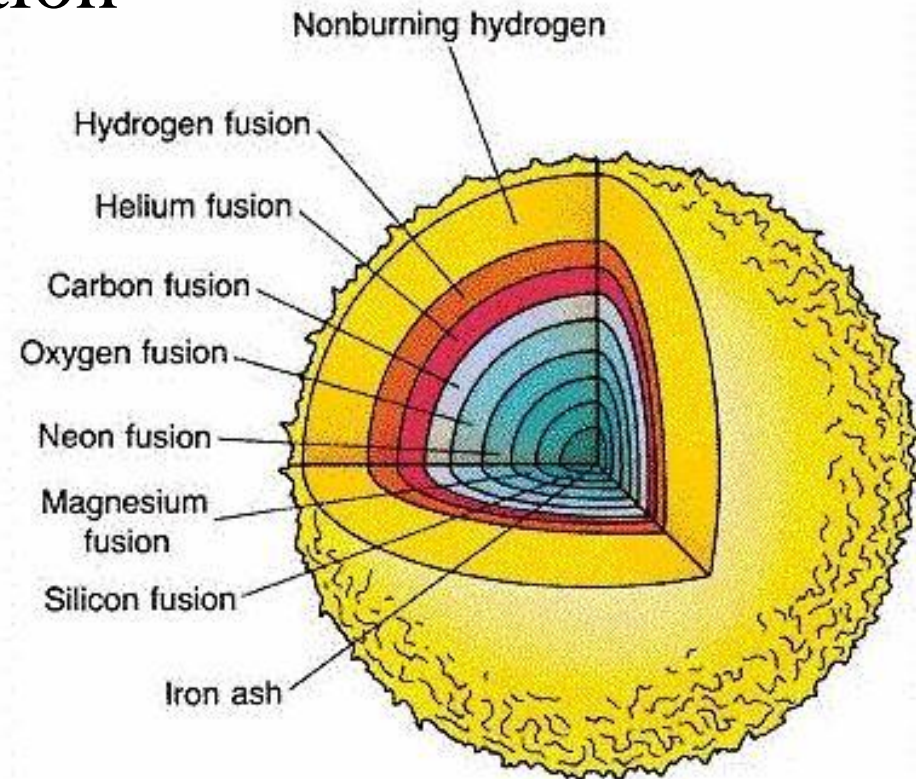
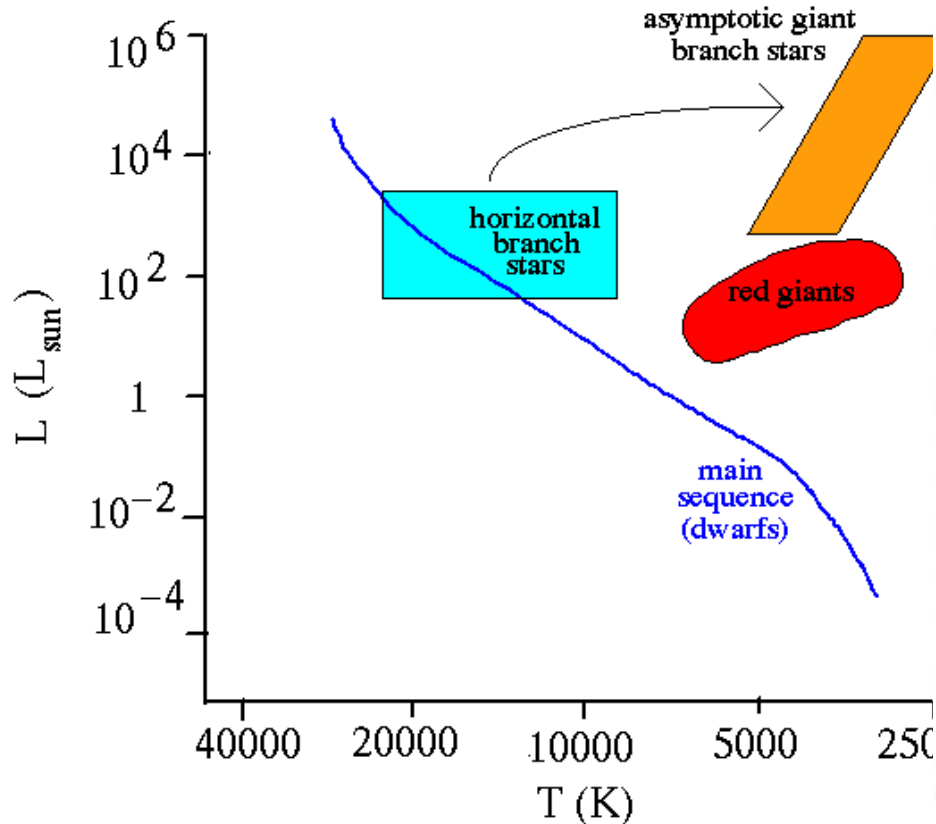


Carbon in the Universe
Miller/Urey experiment
Murchison meteorite
Pages 16-21 in RGS

Axel Brandenburg
(Office hours: Mondays 2:30 – 3:30 in X590
and Wednesdays 11-12 in D230)

Organic matter in the Universe

- Carbon comes from nuclear fusion in stars
- Late stages of evolution



an "old" sun

Stellar evolution

- Stars consume food (H)
- Expell waste (light, gas, dust)
- Grow (protostar, main sequence, giant)
- Die, and produce offspring from their dust

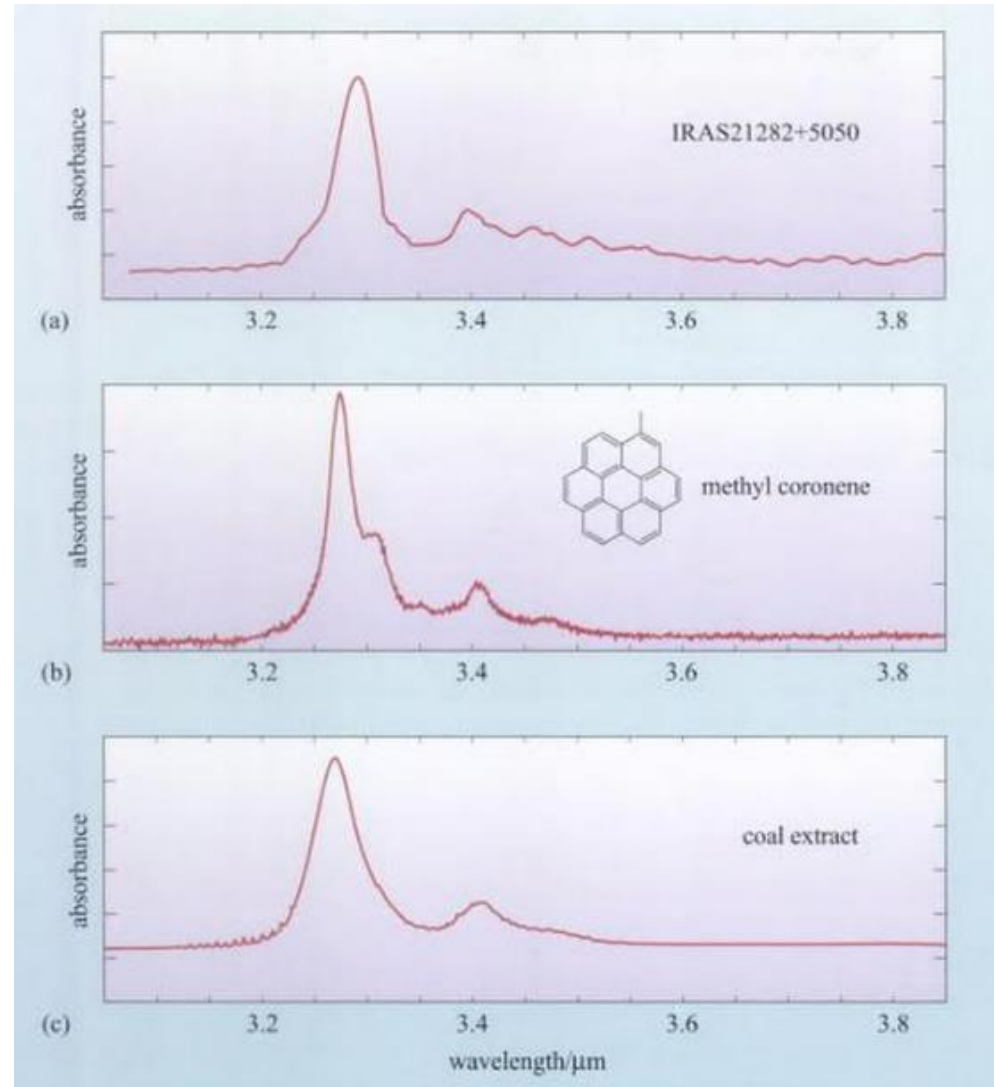
...are stars alive?

Stellar “evolution”

- No Darwinian evolution
- This “life” cycle is more like a motor
- But let’s now look at their waste...

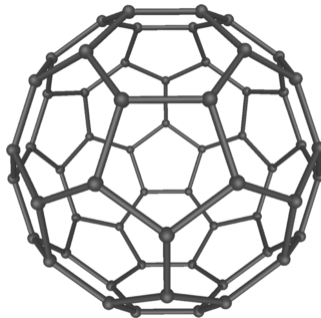
Similarity of spectra with coal

- Circumstellar envelope of IRAS21282+5050
- Aromatic carbons *common* in space
- Coal – the very bottom of bottom-up approach



Molecules around stars

- $\boxed{\text{H}_2}$ smallest
- CO, CO_2
- HCOOH
- $\text{CH}_3\text{-CH}_2\text{-OH}$
- NH_3, HCN
- $\text{H}_2\text{S}, \text{N}_2\text{O}$
- $\text{C}_{60}, \boxed{\text{C}_{70}}$ largest



hydrogen species			
H_2	HD	H_3^+	H_2D^+
hydrogen and carbon compounds			
CH	CH^+	C_2	CH_2
CH_3	C_2H_2	$\text{C}_3\text{H (lin)}$	$\text{C}_3\text{H (circ)}$
$\text{H}_2\text{CCC (lin)}$	C_4H	C_5	C_2H_4
$\text{CH}_3\text{C}_2\text{H}$	C_6H	H_2C_6	C_7H
hydrogen, carbon and oxygen compounds			
OH	CO	CO^+	H_2O
HOC^+	C_2O	CO_2	H_3O^+
C_3O	CH_2CO	HCOOH	H_2COH^+
C_5O	CH_3CHO	$\text{C}_2\text{H}_4\text{O (circ)}$	CH_3OCHO
CH_3OCH_3	$\text{CH}_3\text{CH}_2\text{OH}$	$(\text{CH}_3)_2\text{CO}$	
hydrogen, carbon and nitrogen compounds			
NH	CN	NH_2	HCN
NH_3	HCNH^+	H_2CN	HCCN
CH_2NH	HC_2CN	HC_2NC	NH_2CN
CH_3NC	HC_3NH^+	C_5N	CH_3NH_2
$\text{CH}_3\text{C}_3\text{N}$	$\text{CH}_3\text{CH}_2\text{CN}$	HC_7N	$\text{CH}_3\text{C}_5\text{N}$
hydrogen, carbon (possibly), nitrogen and oxygen compounds			
NO	HNO	N_2O	HNCO
other species			
SH	CS	SO	SO^+
SiC	SiN	SiO	SiS
AlCl	KCl	HF	AlF
H_2S	C_2S	SO_2	OCS
NaCN	MgCN	MgNC	H_2CS
HSiC_2	SiC_3	SiH_4	SiC_4

Darwin's early ideas

(letter to Hooker of 1871)

- ”... we could conceive in some warm little pond, with all sorts of ammonia and phosphoric salts, light, heat, electricity, etc, present that a protein compound was chemically formed ready to undergo still more complex changes”

Notes of music

- Carl Sagan:
 - Building blocks like notes to the music
 - Not the music itself
- Ramming right molecules together
 - as unlikely as hearing Beethoven's 9th when letting monkeys loose with roomfull of instruments

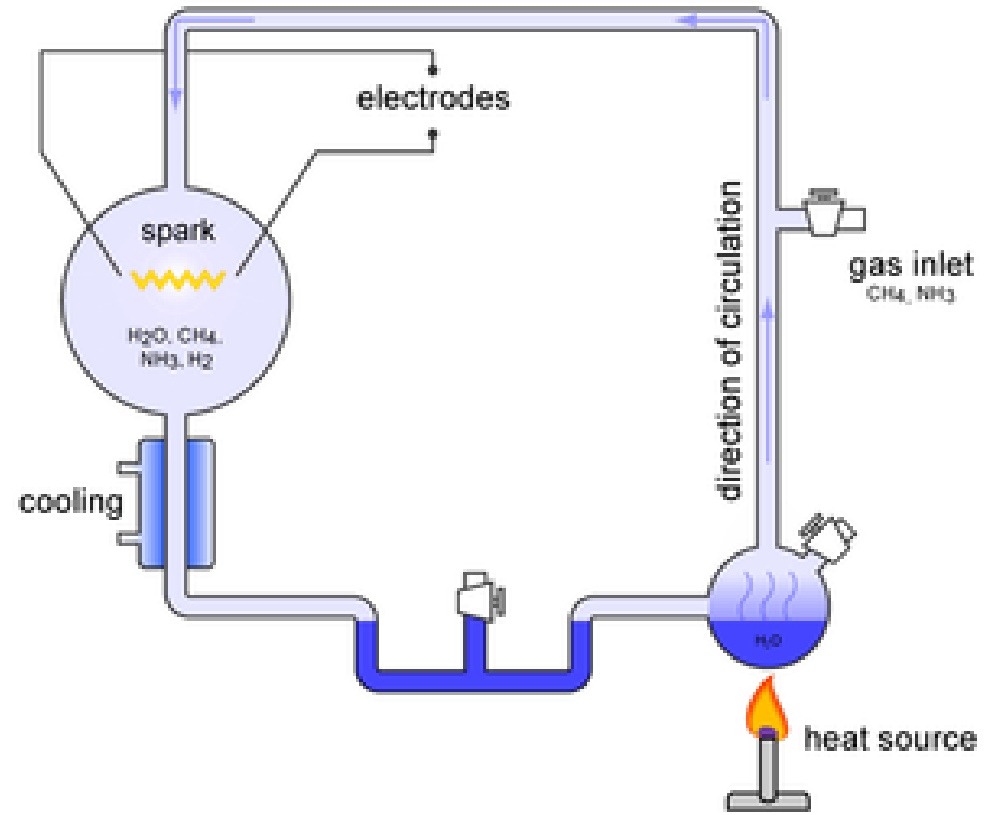
Putting the right molecules together

- Need to produce order
 - drive away from equilibrium
- Energy required to generate & sustain order

Source	Energy /J m ⁻² yr ⁻¹
total radiation from the Sun	1 090 000.0
ultraviolet light	1 680.0
→ electric discharges (lightning)	1.68
cosmic rays	0.000 6
radioactivity (to 1 km depth)	0.33
volcanoes	0.05
shock waves (atmospheric entry)	0.46

Comparison with Frankenstein experiment

- Reducing atmosphere:
- H_2 , H_2O ,
 NH_3 , CH_4 ,
Oparin/Haldane
- Discharges
- Steam



After 1 week....

- Brown substance
- Glycine, alanine, (valine, proline, ...)

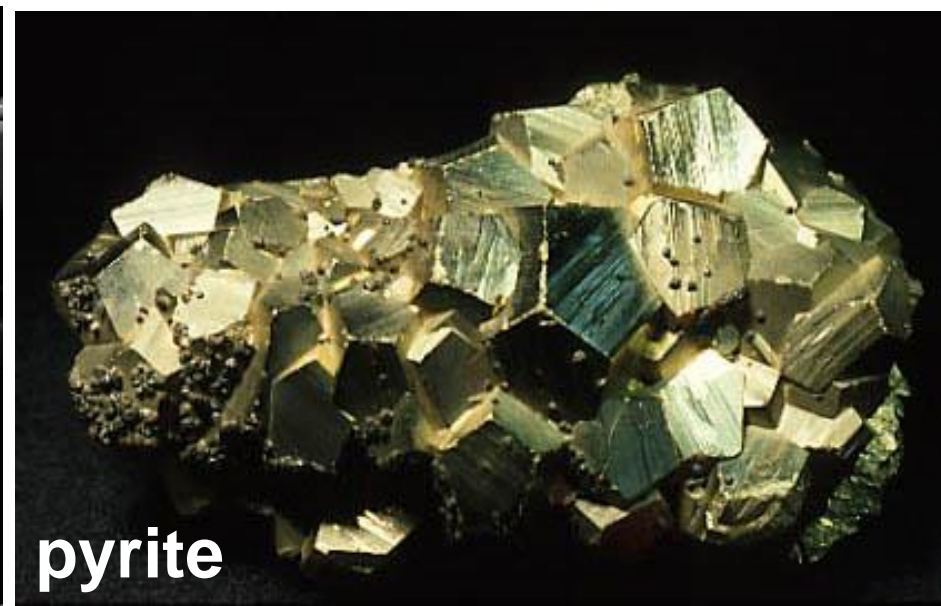
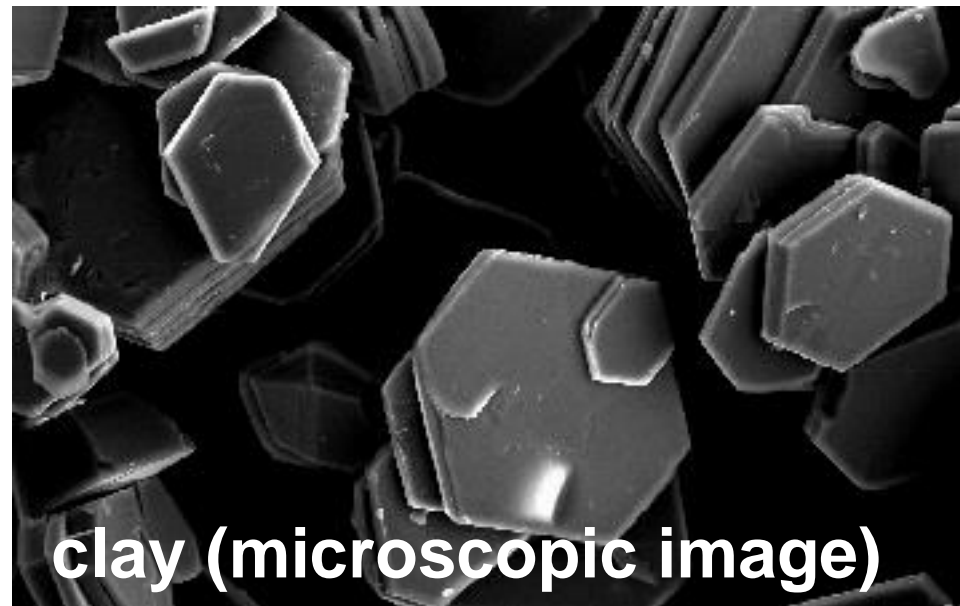


Role of minerals

- Support
 - Amino acids polymerize on surfaces
- Selection
 - Different crystal faces select left/right
 - Both possible → natural selection chose one
- Catalysis
 - N_2 to N_3H via metallic surfaces
 - Suitable in hydrothermal vents

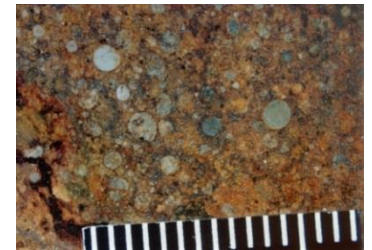
Minerals as Templates?

- How could first RNA “genetic” strands arise?
- Lab experiments: clays + water + organics \Rightarrow complex organics, including RNA strands.
- Repeating chemical structure of minerals may facilitate assembly of complex organics!



Meteorites: organics from space

- Stony:
 - Chondrites: 86%
 - Ordinary (80% of those)
 - Carbonateous (5% of those)
 -
 - Achondrites: 8%
- Iron: 6%

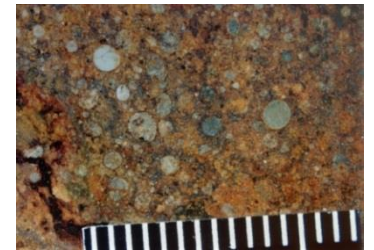


Chelyabinsk
(2013)

1500 people injured!

Meteorites: different types

- Stony:
 - Chondrites: 86%
 - Ordinary (80% of those)
 - Carbonateous (5% of those)
 -
 - Achondrites: 8%
- Iron: 6%



Murchison meteorite

- September 28, 1969, 10:58 am, Melbourne
 - Observed as fireball, w/ smoke for 30s
 - Broke up into pieces, altogether >100kg
- Carbonaceous chondrite
 - Smelly when found
- Analyzed like moon rocks
 - >15 amino acids, etc



Comparison

Amino acid	Abundance of amino acids	
	synthesized in the Miller-Urey experiment	Found in the Murchison meteorite
glycine	●●●●	●●●●
alanine	●●●●	●●●●
α -amino- <i>N</i> -butyric acid	●●●	●●●●
α -aminoisobutyric acid	●●●●	●●
valine	●●●	●●
norvaline X	●●●	●●●
isovaline X	●●	●●
proline	●●●	●
pipecolic acid X	●	●
aspartic acid	●●●	●●●
glutamic acid	●●●	●●●
β -alanine	●●	●●
β -amino- <i>N</i> -butyric acid X	●●	●●
β -aminoisobutyric acid X	●	●
γ -aminobutyric acid X	●	●●
sarcosine X	●●	●●●
<i>N</i> -ethylglycine X	●●	●●
<i>N</i> -methylalanine X	●●	●●

not used in life

Organics in Murchison & Life

	Role	Life	Murchison meteorite
water	solvent	yes	yes
lipids (hydrocarbons and acids)	membranes, energy storage	yes	yes
sugars (monosaccharides)	} support, energy storage	yes	yes
polysaccharides (polymerized sugars)		yes	no
amino acids	} many (support, enzymes, etc.)	yes	yes
proteins (polymerized amino acids)		yes	no
phosphate		yes	yes
nitrogenous bases	} genetic information	yes	yes
nucleic acids (polymerized sugars, phosphates and nitrogenous bases)		yes	no

- Long polymers absent (single sugars, no proteins)
- No nucleotide nor nucleic acids

*Monomers of building blocks: ...what is **not** found*

- A. amino acids
- B. sugars
- C. nucleotides
- D. lipids
- E. none (all are found)

Monomers of building blocks

...what is not found

- A. amino acids
- B. sugars
- C. nucleotides
- D. lipids
- E. none (all are found)

Organics in Murchison & Life

	Role	Life	Murchison meteorite	
	water	solvent	yes	yes
X	lipids (hydrocarbons and acids)	membranes, energy storage	yes	yes
X	sugars (monosaccharides)	support, energy storage	yes	yes
	polysaccharides (polymerized sugars)		yes	no
X	amino acids	many (support, enzymes, etc.)	yes	yes
	proteins (polymerized amino acids)		yes	no
	phosphate		yes	yes
	nitrogenous bases	genetic information	yes	yes
X	nucleic acids (polymerized sugars, phosphates and nitrogenous bases)		yes	no

- Long polymers absent (single sugars, no proteins)
- No nucleotide nor nucleic acids

X: building block of life (→ all present in Murchison!)

The importance of the Miller/Urey experiment is

- A. It proved beyond doubt that life could have arisen naturally on Earth;
- B. it showed that natural chemical reactions can produce building blocks for life;
- C. it showed that clay can catalyze the production of RNA.

The importance of the Miller/Urey experiment is

- A. It proved beyond doubt that life could have arisen naturally on Earth;
- B. it showed that natural chemical reactions can produce building blocks for life;**
- C. it showed that clay can catalyze the production of RNA.

Check by elimination

- proved beyond doubt that life could have arisen naturally on Earth?
 - There was no life in this experiment
 - Just amino acids and other organics
- showed that clay can catalyze the production of RNA?
 - Clays were not involved here

The importance of the Miller/Urey experiment is

- A. It proved beyond doubt that life could have arisen naturally on Earth;
- B. it showed that natural chemical reactions can produce building blocks for life;**
- C. it showed that clay can catalyze the production of RNA.

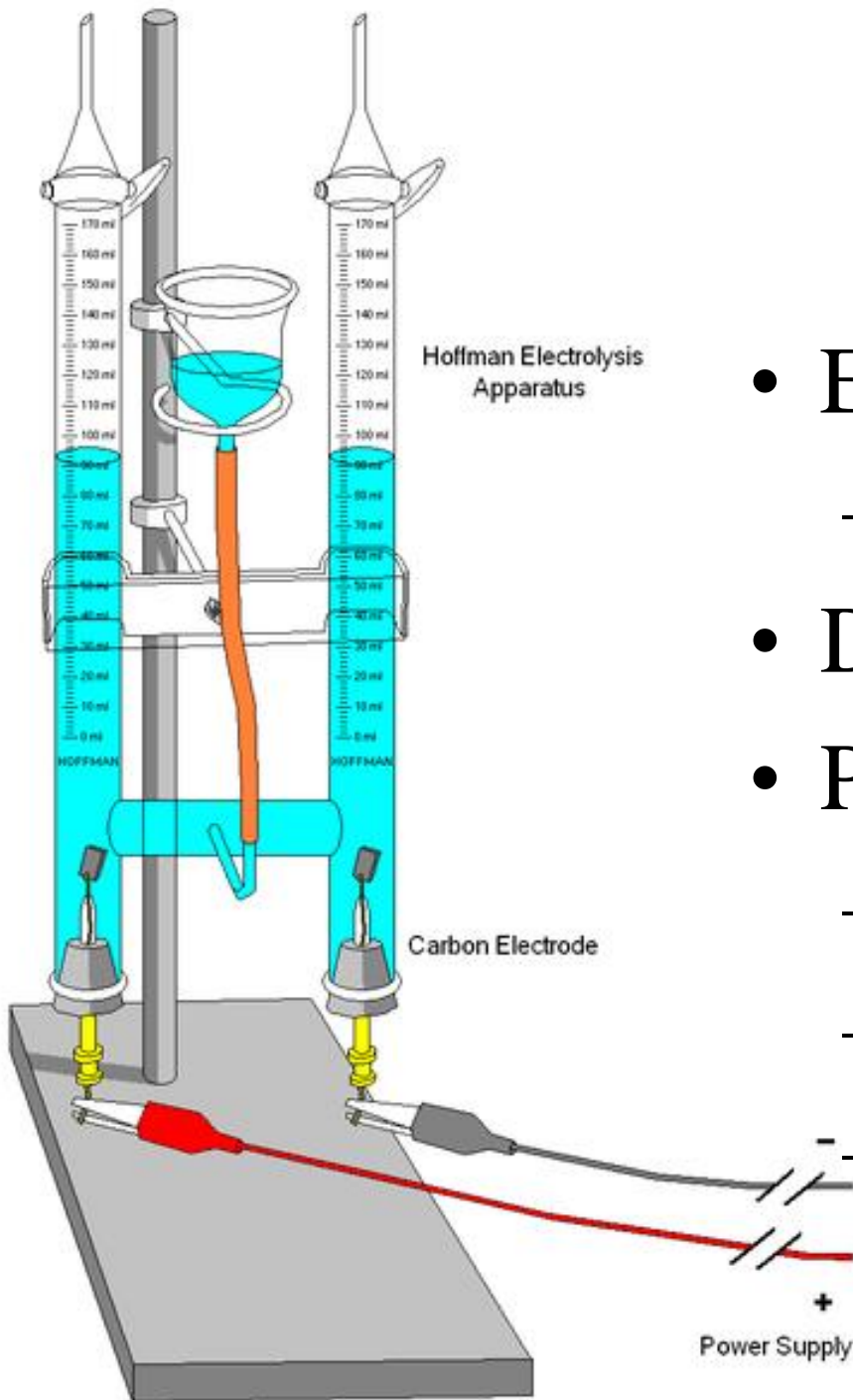
What this really tells us:

- Making many of the building blocks is easy
 - delived from space
 - early atmospheric ?reducing conditions
 - Deep-sea vents (discussed later)
 - Heat & pressure from impacts
- Many possibilities ← no conclusive answ.
- Unclear how to assemble them

What we talked about

- Carbon produced in stars!
- Planetary system made of star dust
- Meteorites deliver amino acids and some of the bases of DNA
- Similar to Miller/Urey experiment
- Organics easy to make → no proofs they have to come from space

Next time



- Electrolysis experiment
 - $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$
- Delivery by comets
- Panspermia Reading:
 - RGS pp. 22-24, 28-29
 - Lon pp. 383-384
 - BS pp. 121-125